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Surface Roughness Of Teeth After Application Of Rematerializing Toothpastes- An In Vitro Study

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Abstract

The aim of this study reduction of surface roughness of teeth after application of Fluoridated, Calcium Sucrose Phosphate and CPP-ACP Toothpastes.

Materials and Method

The specimens were mounted on acrylic blocks and placed in demineralising solution for 72 hrs. After 72 hrs, initial surface roughness was assessed. The teeth specimens were divided into three groups: Group A: Fluoridated toothpastes Group B: calcium sucrose phosphate toothpaste Group C: CPP-ACP tooth paste. The toothpastes were applied on the specimens for 3 mins and washed with distilled water and placed in artificial saliva. This routine was carried out for 30 days. Final surface roughness was assessed.

Results

In Group 1 (Fluoride), the distribution of mean initial surface roughness did not differ significantly compared to mean final surface roughness. In Group 2 (CSP), the distribution of mean initial surface roughness is significantly higher compared to mean final surface roughness. In Group 3 (CPP-ACP), the distribution of mean initial surface roughness is significantly/ higher compared to mean final surface roughness (P-value<0.001).

Conclusion

Within the limitations of this study, CPP-ACP and calcium sucrose phosphate showed the significant reduction in the surface roughness compared to fluoridated toothpaste.

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Keywords

CPP-ACP, Calcium Sucrose Phosphate, Dental Caries, Surface Roughness

Introduction

Dental caries is most prevalent chronic disease affecting the human race. Its prevalence has been noted since the time of pre-Neolithic humans. With respect to all the definitions by various authors, dental caries is initiated via demineralisation of the dental hard tissues resulting in roughing of the hard tooth surface Caries results from an imbalance between many cycles of demineralization and remineralization rather than from continued demineralization. The earliest clinical sign is the "white spot lesion" which result in roughening of the surface of teeth.

Specimen Preparation

After obtaining the Institute Ethical clearance, the study was conducted in the Department of Conservative Dentistry and Endodontics. Sixty human extracted single-rooted teeth with single canal and well-formed root apices, were collected from Department of Oral and Maxillofacial Surgery with informed consent of the donor.

Preparation of Demineralization solution

Dematerializing solution was prepared in the Department of Biochemistry. A digital pH meter (Slope Labtronics, Model LT-11, Punchkula, Haryana, India) was used to check pH during and after preparation of solution.

Preparation of Artificial Saliva

We made the artificial saliva we used in our study according to the formulation used by McKnight Hanes and White ford which contained 0.65 grams per liter potassium chloride 0.058 g/L magnesium chloride BP, 0.165 g/L calcium chloride BP, 0.804 g/L dipotassium hydrogen phosphate, 0.365 g/L potassium dehydrogenate phosphate, 2 g/L sodium benzoate, 7.8 g/L sodium carboxymethyl cellulose BP, deionized water to make 1 L.

Specimen Treatment

The acrylic blocks were placed in artificially prepared deminerazation solution using Standard Ph cycle. A daily schedule of 3 cycles of 30 mins of demineralisation and 2.5 hrs of remineralization was performed after which the specimens were placed in artificial saliva for 72 hrs. The specimens were then divided into 3 groups:

Group A: Specimens with application of fluoridated (Colgate, Colgate-Palmolive, India) toothpaste

Group B: Specimens with application of Calcium sucrose phosphate (Tooth in, Abbott India Ltd.) toothpaste

Group C: Specimens with application of CPP-ACP toothpaste.

Initial Surface roughness was analysed using surface profilometry prior to the application of the toothpaste. Toothpastes were applied on the crown disks using a micro brush applicator tip and tooth brushing was performed using a micro motor and bristle brush for about 3 mins.

Discussion

It marks the beginning of early enamel caries. During demineralization, Ca2+, OH–, PO4 2–, F–, CO3–, Na+ and Mg2+ get displaced from the enamel surface to the exterior. More the acidic environment, greater is the outward flow of ions. However, mineral content of surface is higher than the body of the lesion.^[11] Thus, pH plays a major role in the loss of these ions resulting in surface roughness.

Conclusion

Within the limitations of this study, CPP-ACP and calcium sucrose phosphate showed the significant

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reduction in the surface roughness compared to fluoridated toothpaste.

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